Environmentally-friendly superhydrophobic and superoleophobic fabrics prepared from water-based suspensions

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Avec le soutien du Fonds européen de Développement Régional







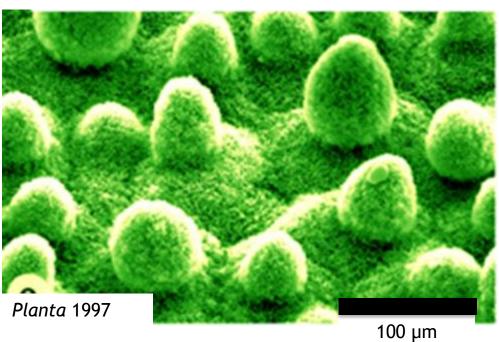




Motivation - Super-water-repellency

Lotus leaf





Motivation - Super-water-repellency

Lotus leaf



Anti-staining clothing



UltraTech International

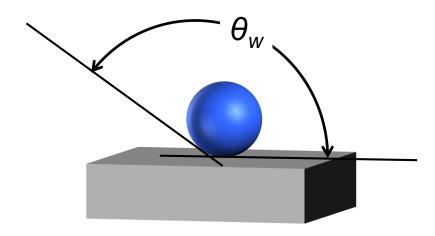
Outline

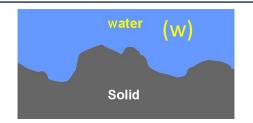
- Theoretical background
 - Wenzel Model and Cassie-Baxter Model
 - Parameters to control wetting behaviors
- Results
 - Fabrics treated by a water-based suspension of C4 fluorinated polymer + silica particles
 - Fabrics treated by water-based silicone rubber and silica particle suspensions
- Conclusions

Wenzel's model

$$\cos\theta_w = R \cos\theta_0$$

 θ_0 : contact angle on an ideal flat surface depending on surface chemistry



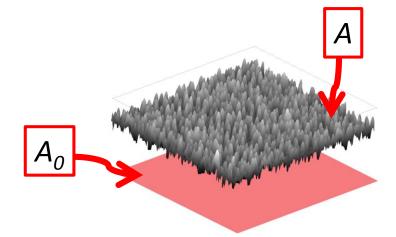


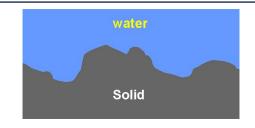
Wenzel's model

$$\cos\theta_w = R \cos\theta_0$$

 θ_0 : contact angle on an ideal flat surface depending on surface chemistry

R: surface roughness





$$R = \frac{A}{A_0}$$

Wenzel's model

$$\cos\theta_w = R \cos\theta_0$$

 θ_0 : contact angle on an ideal flat surface depending on surface chemistry

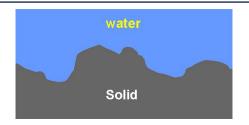
R : surface roughness

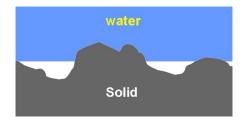


$$\cos\theta_w = f_s (R \cos\theta_0 + 1) - 1$$

 f_s : surface fraction of solid

 $1-f_s$: surface fraction of air-filled cavities





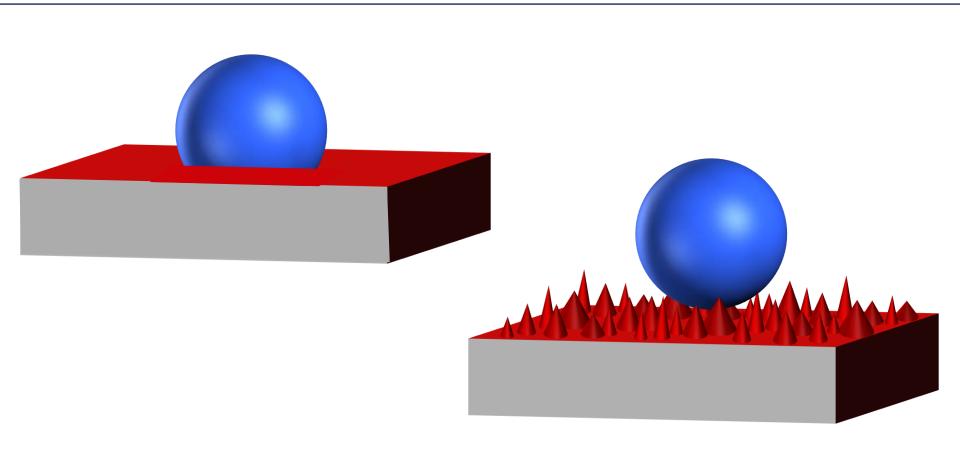
Two general parameters to control wetting behaviors:

- Surface roughness/structure: R and f_s
- Surface chemistry: θ_0

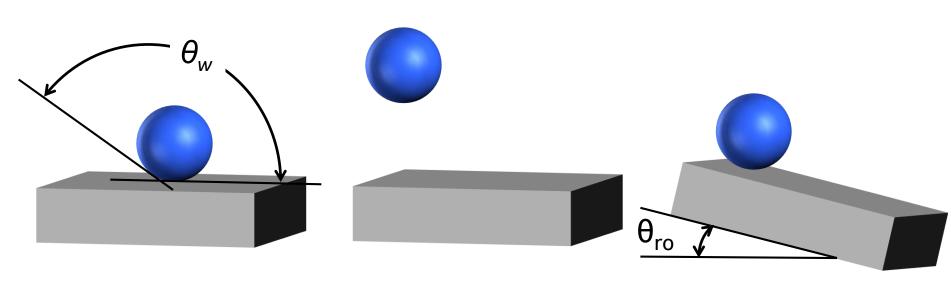
No.1:

Surface roughness/structure

Surface roughness is one parameter controlling the contact angle



Surfaces of sufficiently high roughness may become superhydrophobic



Water contact angle $\theta_w > 150^\circ$

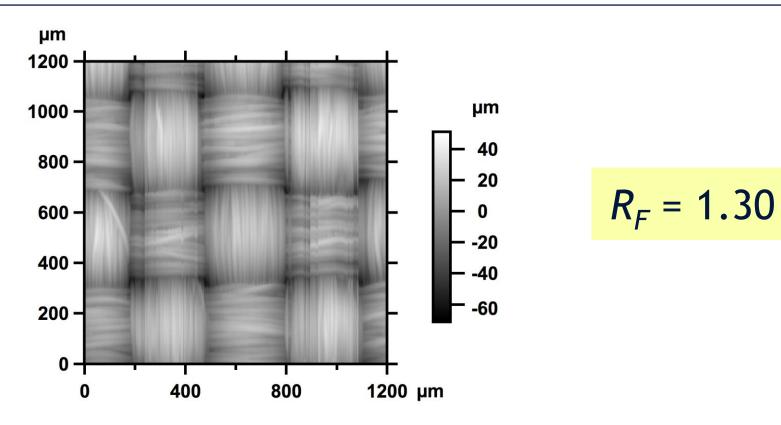
Droplet bouncing

Droplet roll-off $\theta_{ro} < 5^{\circ}$

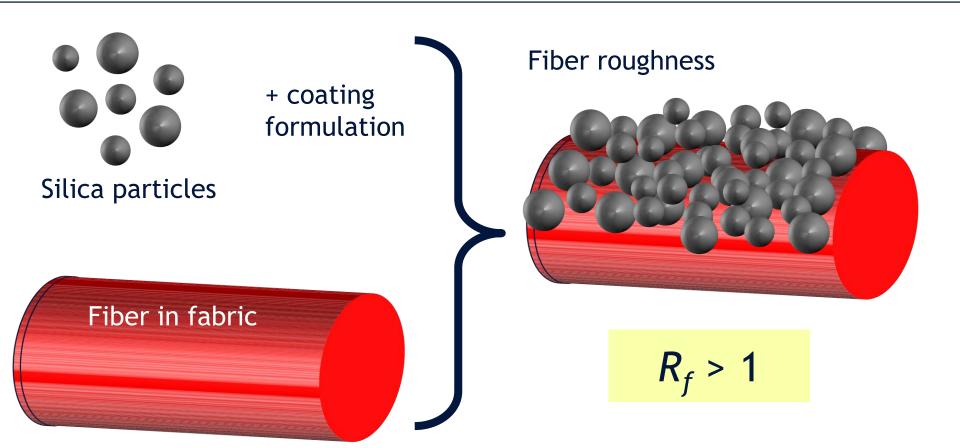
What does contribute

to the fabric roughness?

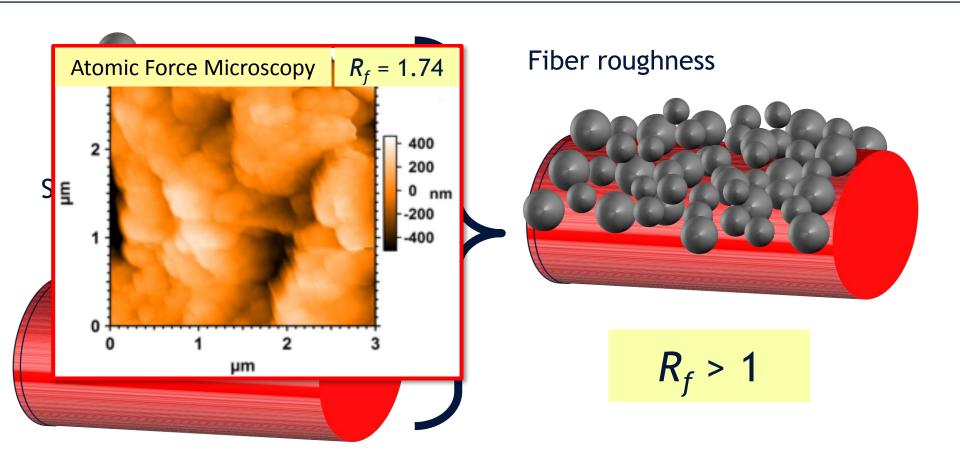
Woven fabric has an intrinsic roughness (R_F) which can be measured by profilometry



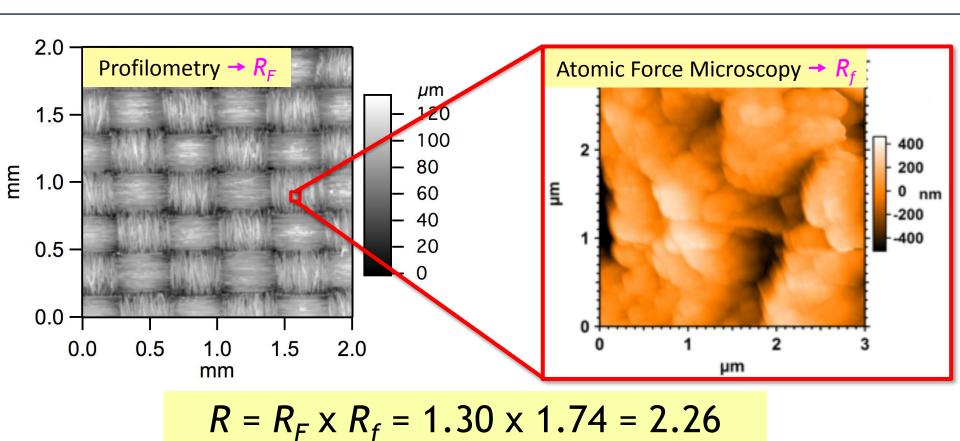
Boosting fiber roughness (R_f) with silica particles



Boosting fiber roughness with silica particles



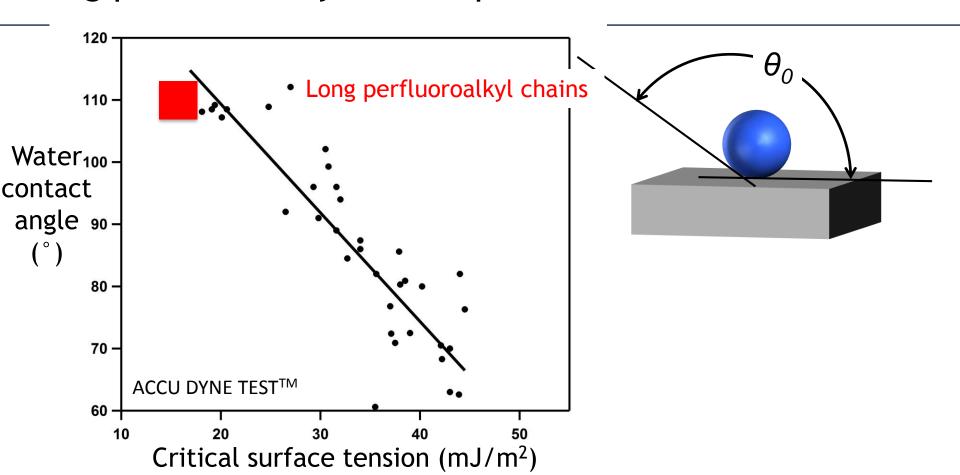
Total fabric roughness (R)



No. 2: Surface Chemistry

Why were long perfluoroalkyl chains used for water repellency?

Long perfluoroalkyl chains provide low surface tension



'Long' (C8) perfluoroalkyl chains are being banned

L 150/14

EN

Official Journal of the European Union

14.6.2017

COMMISSION REGULATION (EU) 2017/1000

of 13 June 2017

amending Annex XVII to Regulation (EC) No 1907/2006 of the European Parliament and of the Council concerning the Registration, Evaluation, Authorisation and Restriction of Chemicals (REACH) as regards perfluorooctanoic acid (PFOA), its salts and PFOA-related substances

(Text with EEA relevance)

THE EUROPEAN COMMISSION,

Having regard to the Treaty on the Functioning of the European Union,

Having regard to Regulation (EC) No 1907/2006 of the European Parliament and of the Council of 18 December 2006 concerning the Registration, Evaluation, Authorisation and Restriction of Chemicals (REACH), establishing a European Chemicals Agency, amending Directive 1999/45/EC and repealing Council Regulation (EEC) No 793/93 and Commission Regulation (EC) No 1488/94 as well as Council Directive 76/769/EEC and Commission Directives 91/155/EEC, 93/67/EEC, 93/105/EC and 2000/21/EC (¹), and in particular Article 68(1) thereof,

'Long' (C8) perfluoroalkyl chains are being banned



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Prioritizing Existing Chemicals for Risk Evaluation

Risk Evaluations for Existing Chemicals Under TSCA

Current Chemical Risk Management Activities

Risk Management for Per- and Polyfluoroalkyl Substances (PFASs) under TSCA

Overview

Background

PFOA Stewardship Program

More Information

Overview

What are PFASs and where are they found?

Many per- and polyfluoroalkyl substances (PFASs), also referred to as perfluorinated chemicals (PFCs), are found world-wide in the environment wildlife and humans

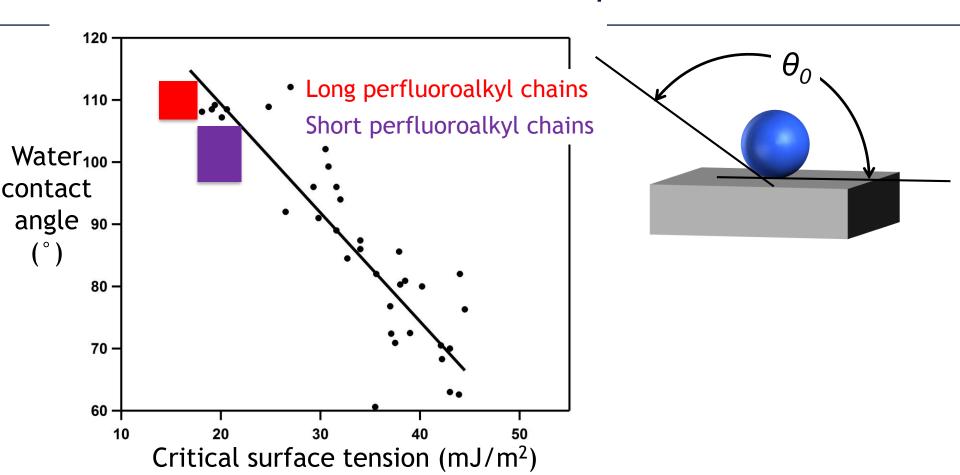
Announcements

On May 19, 2016, EPA

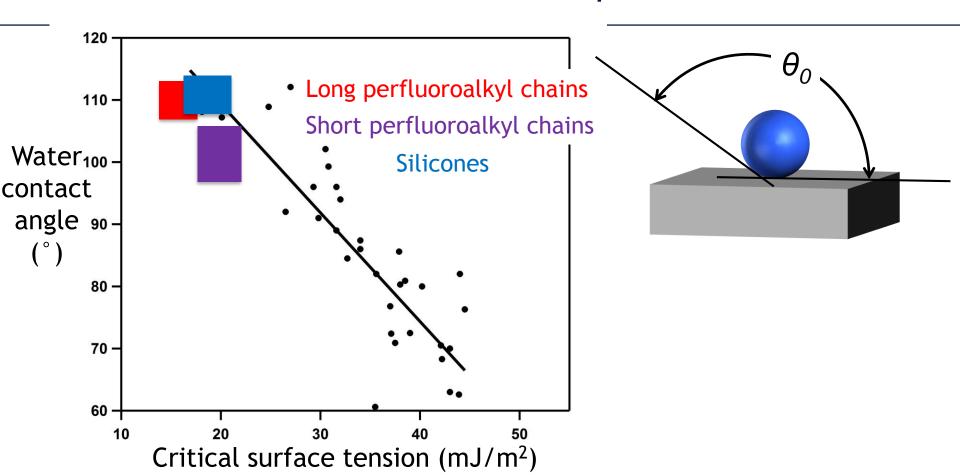
to replace long perfluoroalkyl chains?

Are there possible alternative compounds

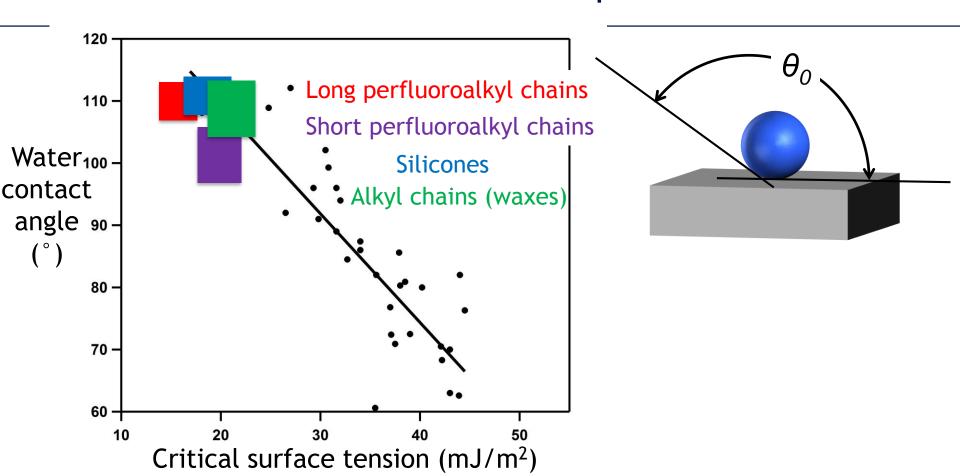
Other candidates are possible



Other candidates are possible



Other candidates are possible



Outline

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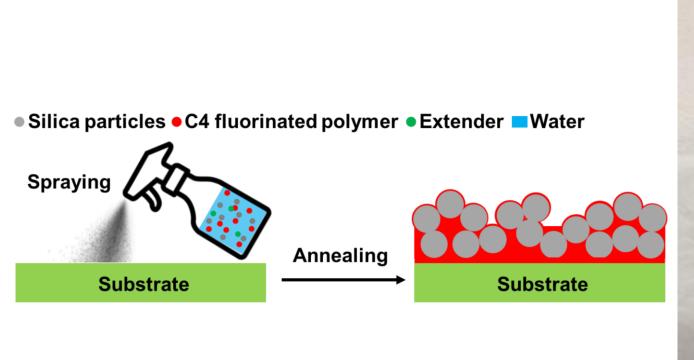
One step deposition of an aqueous suspension containing:

Polyurethane bearing short (C4) perfluoroalkyl chains (*PM900*®, *3M*)

+

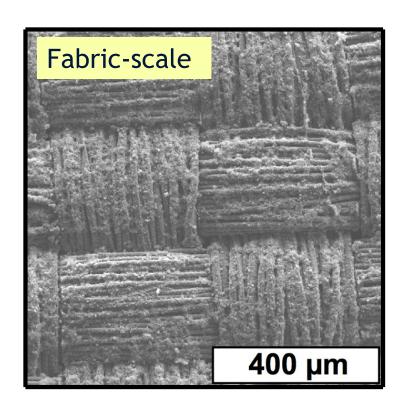
Silica particles (*Tixosil*®, *Solvay*)

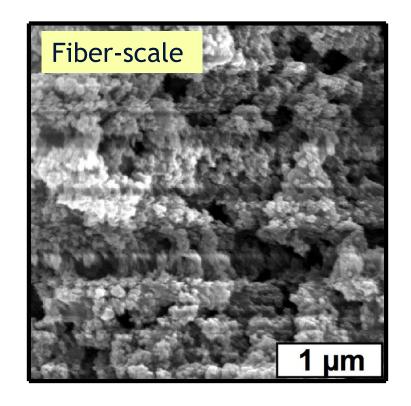
Spraying the one-pot aqueous formulation



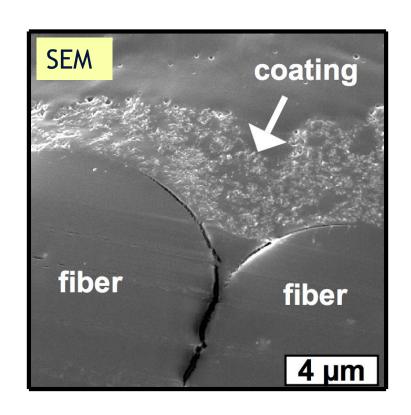


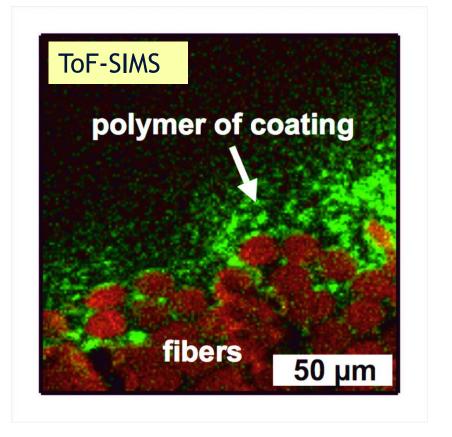
Surface morphology of a typical sample (SEM)

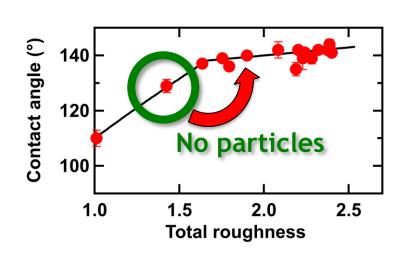


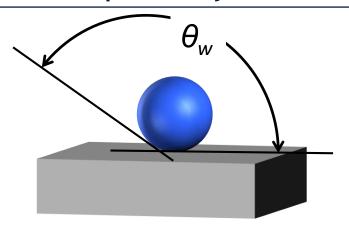


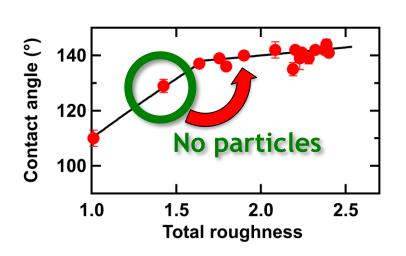
A typical transverse cut

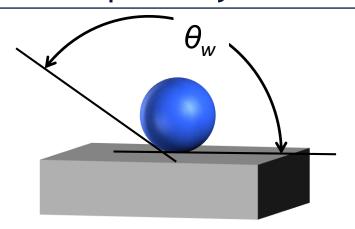




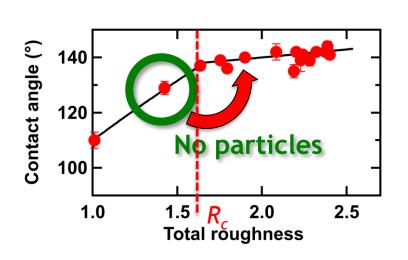


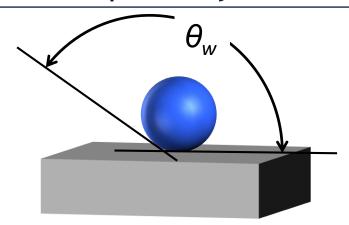






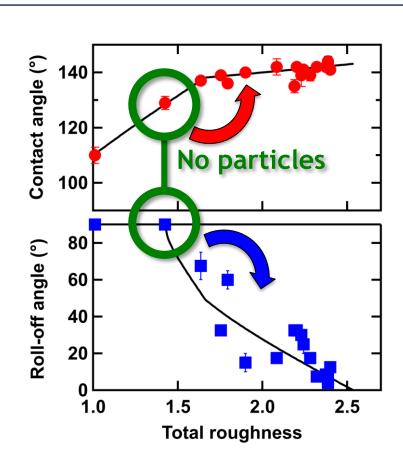
Fitting using Cassie-Baxter equation: $\cos \theta_w = f_s (R \cos \theta_0 + 1) - 1$

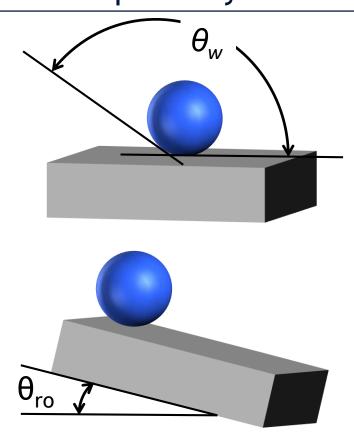




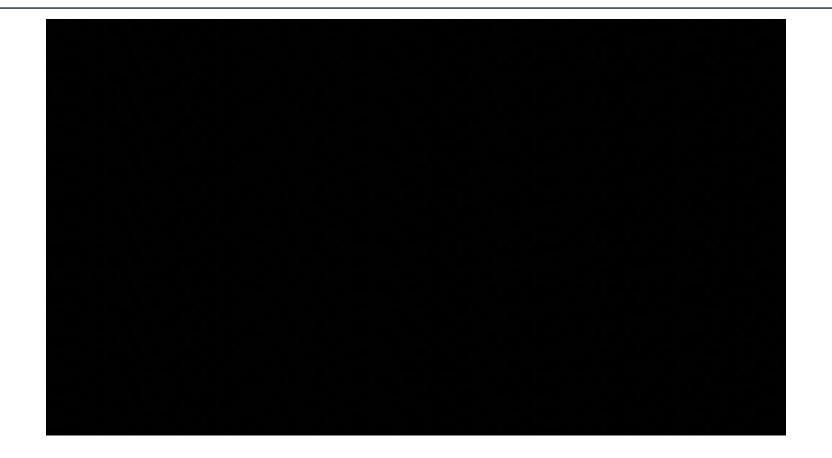
Fitting using Cassie-Baxter equation: $\cos \theta_w = f_s (R \cos \theta_0 + 1) - 1$

 $1-f_s = 0.71$ (fraction of trapped air) $R_c = 1.6$ (critical roughness)





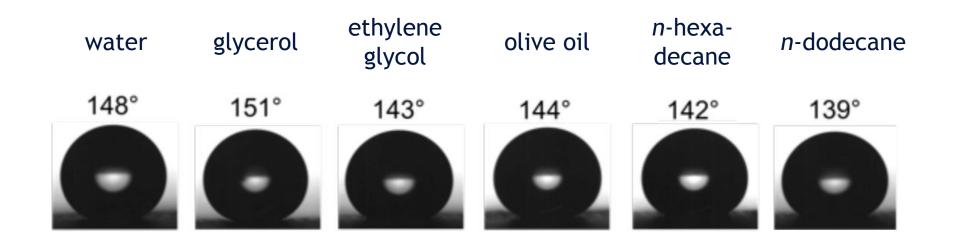
Water roll-off on a superhydrophobic C4-based fabric



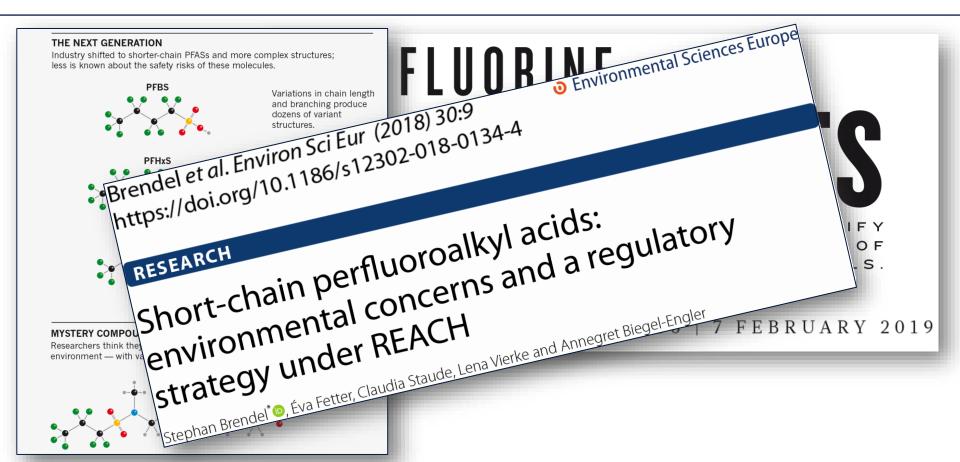
The short C4 perfluoroalkyl chains also lead to superoleophobicity



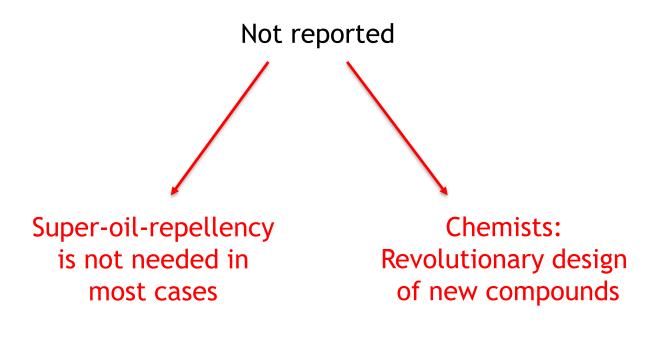
The short C4 perfluoroalkyl chains also provide resistance to some organic solvents



However, shorter perfluoroalkyl chains also raise concerns



Super-oil-repellency without fluorinated compounds



Step-by-step deposition of aqueous suspensions incorporating:

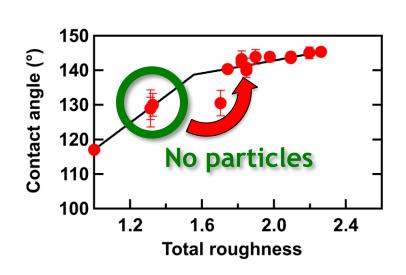
a crosslinked silicone rubber (HC303®, Wacker)

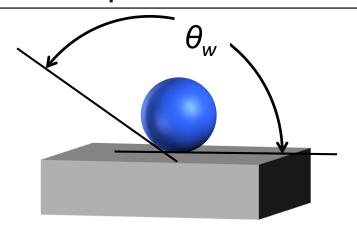
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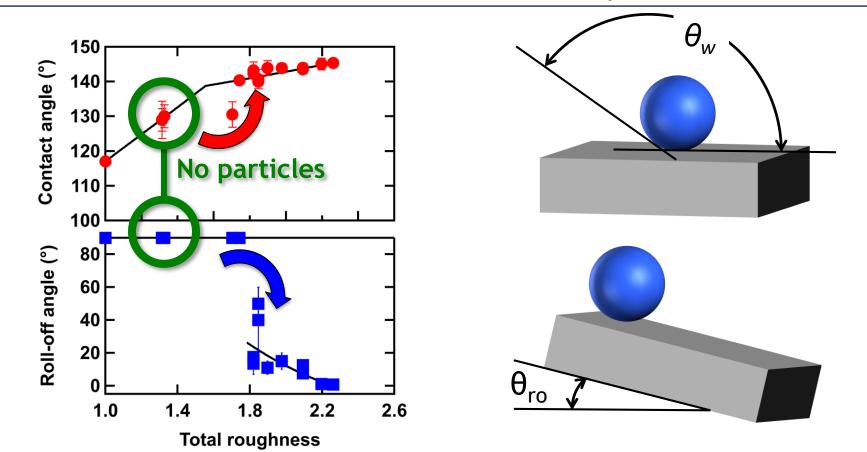
Silica particles (*Tixosil*®, *Solvay*)

Silicone-based superhydrophobic coatings (dip-coating from aqueous suspensions)

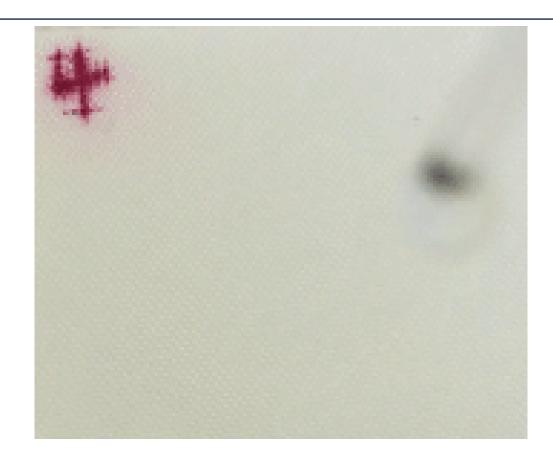




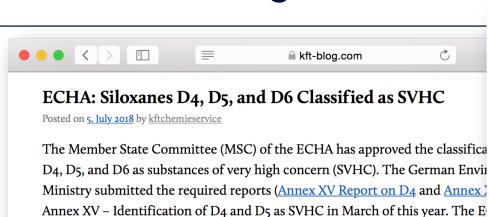




Water roll-off on a silicone-based superhydrophobic fabric



Silicones might also rise concerns in the public



the Report on D6.

The experts in the MSC rated D4 as persistent, bioaccumulative (it accumulate chain), and toxic (PBT), but they assigned only persistent and bioaccumulative to D5 and D6. Nevertheless, D5 and D6 can also be classified as PBT when bot show D4 impurities in a concentration equal to or greater than 0.1% by weight

Silicone compounds D4 (cyclotetrasiloxane), D5 (cyclopentasiloxane), and D6 (cyclohexasiloxane) are often found in personal care products and flow into the along with waste water. They are also important source materials for certain significant found as residues in finished products.

EUROPEAN CHEMICALS AGENCY

Substance Name:

Octamethylcyclotetrasiloxane (D4)

EC Number: 209-136-7

CAS Number: 556-67-2

MEMBER STATE COMMITTEE
SUPPORT DOCUMENT

FOR IDENTIFICATION OF

OCTAMETHYLCYCLOTETRASILOXANE (D4)
AS A SUBSTANCE OF VERY HIGH CONCERN
BECAUSE OF ITS PBT¹ AND vPvB² PROPERTIES

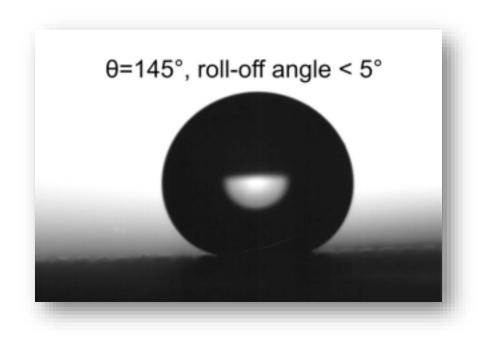
(ARTICLE 57D&E)

Adopted on 13 June 2018

PBT means persistent, bioaccumulative and toxic vPvB means very persistent and very bioaccumulative

Preliminary studies indicate wax-based alternatives to be promising

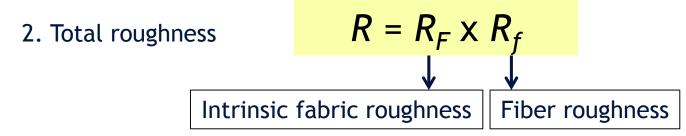
Silicone replaced by water-based paraffin wax (Contraqua WE) (total roughness R: $1.3 \times 1.72 = 2.23$)





Main conclusions

1. Total roughness is a predictor of water repellent performance; other parameters may have to be considered



- 3. Fiber surface roughness can be boosted by silica particles; other methods certainly exist
- 4. Superhydrophobic fabrics can be obtained by using non-fluorinated compounds such as waxes and silicones , in combination with silica particles
- 5. Short perfluoroalkyl chains (C4) can be used to replace long ones to achieve super-water/oil-repellency, in combination with silica particles

Acknowledgments







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